

## METHOD DEVELOPMENT AND VALIDATION OF OMEPRAZOLE IN BULK AND MARKETING FORMULATION BY USING UV-SPECTROPHOTOMETRY

Chinababu D.<sup>1</sup>, Vishal Bharat Babar<sup>1</sup>, Mane Yogita S.<sup>1\*</sup>, Shevale Harshala S.<sup>1</sup>, More Nikita Y.<sup>1</sup>, Kokane Sonali P.<sup>1</sup>, Rakh Shital S.<sup>1</sup>, Aleesha SK<sup>2</sup>

<sup>1</sup>Institute of Pharmaceutical Science and Research for Girls, Bhigwan-413130, Maharashtra.

<sup>2</sup>Dattakala College of Pharmacy, Swami-Chincholi, Bhigwan-413130, Maharashtra.

Article Received: 21 May 2025 | Article Revised: 11 June 2025 | Article Accepted: 02 July 2025

\*Corresponding Author: Mane Yogita S

Institute of Pharmaceutical Science and Research for Girls, Bhigwan-413130, Maharashtra.

DOI: <https://doi.org/10.5281/zenodo.15879705>

**How to cite this Article:** Chinababu D., Vishal Bharat Babar, Mane Yogita S., Shevale Harshala S., More Nikita Y., Kokane Sonali P., Rakh Shital S., Aleesha SK (2025) METHOD DEVELOPMENT AND VALIDATION OF OMEPRAZOLE IN BULK AND MARKETING FORMULATION BY USING UV-SPECTROPHOTOMETRY. World Journal of Pharmaceutical Science and Research, 4(3), 1541-1546. <https://doi.org/10.5281/zenodo.15879705>



Copyright © 2025 Mane Yogita S. | World Journal of Pharmaceutical Science and Research.

This work is licensed under creative Commons Attribution-NonCommercial 4.0 International license (CC BY-NC 4.0)

### ABSTRACT

Omeprazole was estimated using a straightforward, sensitive, and selective UV approach that was devised and verified. The technique was developed using UV spectrometry; the optimal concentration was 8 µg/ml, the absorbance was 0.428, and the  $\lambda_{\text{max}}$  was reached at 301.00 nm. The linearity range of the test technique, which had an  $R^2$  value of 0.998, was 0.5 µg/ml to 16 µg/ml. It was discovered that the precision and accuracy percentage RSD ranged from 0.23 to 1.67. The results indicated that the intraday precision was 0.26 and the interday precision (%RSD) was 0.22. The accuracy (recovery percentage) was found to be within the acceptable range of 99.89–100.21%. The developed method's LOD and LOQ were 0.061 µg/ml and 0.186 µg/ml, respectively. The robustness results were found to be 98%–102%.

**KEYWORDS:** UV-Spectrophotometry, Omeprazole, Method validation, LOD, LOQ.

### INTRODUCTION

The chemical name for omeprazole (OMZ), 5-methoxy-2-[[[4-methoxy-3,5-dimethyl-pyridin-2-yl] methyl]sulfinyl]-1H-benzimidazole (Figure 1), is a proton pump inhibitor that is commonly used as an anti-ulcer medication to treat duodenal and stomach ulcers as well as GERD (gastric reflux disease) and ZERD (Zollinger Ellison syndrome).<sup>[1-3]</sup> It functions by blocking the proton pump. This chemical belongs to the antisecretory class and is a substituted benzimidazole.<sup>[2]</sup> Inhibition of the parietal cell H<sup>+</sup>/K<sup>+</sup> adenosinetriphosphate pump, the final step of acid production.

When taken daily, the effects of omeprazole will level off on the fourth day.<sup>[1-3]</sup> Numerous other methods, including as high-performance liquid chromatography, ultra-performance liquid chromatography, thin layer chromatography, and high-performance thin layer chromatography, have been reported for determining OMZ in bulk and formulations.<sup>[3-24]</sup>

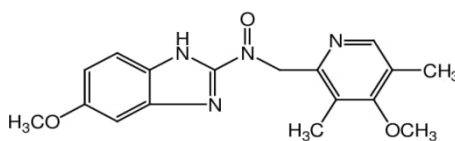


Figure 1: Omeprazole structure<sup>[4]</sup>

## MATERIALS AND METHODS

The reagents and chemicals were all of analytical quality. After double distillation, the water was filtered through a membrane filter. ALSUCROSE Corporation, India, produces ethanol. We purchased Omeprazole, a pharmaceutical-grade standard medication, from MS Pharmaceuticals in Mumbai, Maharashtra, India.

### Preparation of Standard solution

Weighed precisely, 40g of the Omeprazole working standard was deposited into a 50 ml calibrated, clean, and dry volumetric flask. We added roughly 50 ml of solvent (Ethanol: Water), shaken the flask thoroughly, and sonicated it (primary stock solution 800µg/ml) for improved solubility. Using a pipette, remove 0.1 ml of the primary stock solution from the aforementioned solution, then transfer it to a second 10 ml volumetric flask to create the secondary stock solution (8µg/ml).

### Preparation of sample solution

Twenty tablets were precisely weighed, their average weight determined, and they were ground up using a clean motor and pestle. 207 mg of omeprazole tablet powder should be weighed. Then, 40 mg of the corresponding weight should be transferred into a 50 ml volumetric flask. After adding roughly 50 ml of the solvent (ethanol: water), sonicate it until it dissolves fully, then filter as necessary, then replenish the final volume (800µg/ml). Pipette out 0.1 ml of the primary stock solution (above) and pour it into a second 10 ml volumetric flask. Fill it up to the mark to create the secondary stock solution (8µg/ml).

### Optimized parameters of method

The new method's optimal concentration was 8µg/ml and its absorbance was 0.427 and it was optimized at 301 nm using a solvent Water: Ethanol 5:5 (V/V).

### Validation of analytical Method<sup>[25 & 26]</sup>

The suggested approach was verified for a number of criteria, including assay, linearity, precision, accuracy, specificity, robustness, stability studies, LOD, and LOQ.

### Specificity and Selectivity

The response of the spectra to solvent (blank), standard, and sample solutions made using the suggested method was examined in order to investigate the specificity. No interference between the medication and the solvents or excipients was seen. Thus, it was demonstrated that the new approach was both selective and specific. (Figure 2).

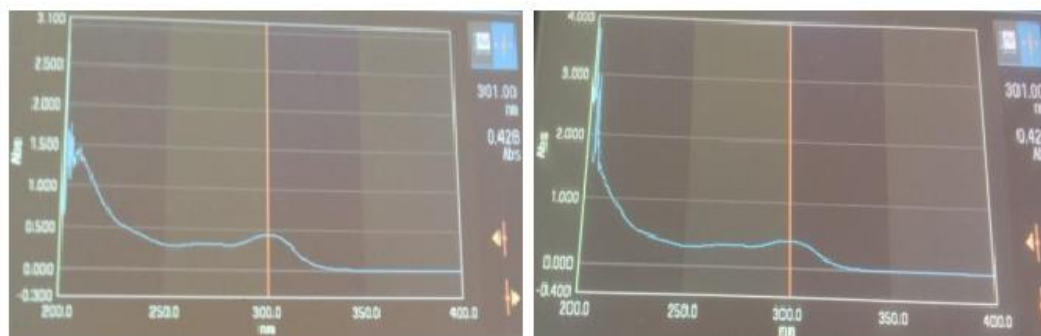


Figure 2: Standard and Sample spectrums of Omeprazole.

### Linearity

The concentration and absorbance response were plotted on a graph (Figure-3), the regression coefficient ( $R^2$ ) was determined to be 0.998. The Omeprazole calibration curve was linear throughout the concentration range. The standard solution concentration was determined to have a linearity range of 0.5  $\mu\text{g/mL}$  to 16  $\mu\text{g/mL}$ . The figure 3 displayed the linearity graph.

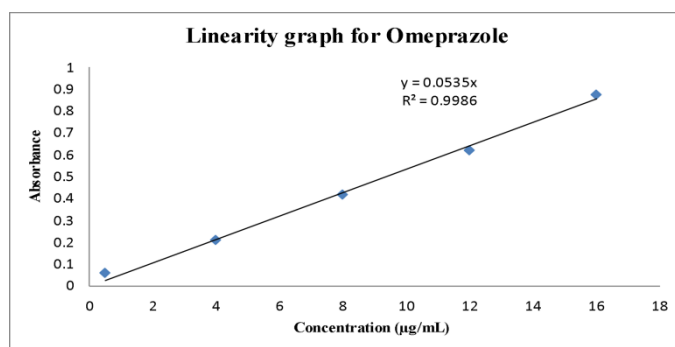


Figure 3: Linearity graph of Omeprazole.

### Precision

Each level of precision, including intraday and intermediate precision, was evaluated using six repeated Omeprazole sample solutions.

#### Intraday Precision

The intraday precision test was conducted three times a day at 9:00 am, 1:00 pm, and 5:00 pm using a sample solution containing 8  $\mu\text{g/mL}$ . Six duplicate measurements were taken at each level. For every interval, the % RSD was determined to be between 0.24 and 0.32.

#### Interday Precision

Using a solution concentration of 8  $\mu\text{g/mL}$ , the interday precision was carried out on days 1, 2, and 3. Six repeated injections' spectrums were recorded at each level, and the average percentage RSD was determined to be between 0.17 and 0.28.

### Accuracy

The accuracy of the process was examined by spiking the standard solution with the analyzed sample solution at three concentration levels: 80%, 100%, and 120%. The recovery experiments were conducted under optimal conditions in

duplicate. The range of accuracy should be 98% to 102%. The RSD percentage value shouldn't be more than 2.0. The results were reported in table 1.

**Table 1: Results of Accuracy.**

Spiked level	Sample weight (mg)	Absorbance	µg/ml added	µg/ml found	% Mean recovery
80%	165.6	0.340	6.36	6.36	100.21
100%	207	0.426	7.95	7.95	100
120%	248.4	0.511	9.54	9.55	99.89

#### Detection limit and quantification limit

The linearity curve of slope and the response of the standard deviation (precision) were used to establish the detection limit and quantification limit. Omeprazole's LOD and LOQ were found to be 0.061 µg/mL and 0.186 µg/mL, respectively.

#### Robustness

The robustness was tested at a concentration of 8 µg/mL with only slight modifications to the method's flow rate and mobile phase composition. The organic phase composition was altered by  $\pm 0.3$  mL, and the wavelength was altered by  $\pm 2$  nm. captured six replicate samples' spectrums. (Table-2).

**Table 2: Results of Robustness.**

Sr. No	Parameter	Condition	Absorbance	% Assay
1	Wavelength ( $\pm 2$ nm)	299	0.424	99.06
2		301	0.428	100.00
3		303	0.425	99.29
4	Changed Organic Solvent ratio ( $\pm 0.3$ mL)	5.3:4.7	0.429	100.94
5		5:5	0.426	99.53
6		4.7:5.3	0.423	98.83

#### CONCLUSION

The development and validation of various UV-Spectrophotometric techniques for the measurement of omeprazole in pharmaceutical dosage forms and bulk was attempted. It was discovered that the suggested spectrophotometric approach was straightforward, accurate, and precise. The technique was created using a 5:5 V/V ratio of ethanol to water. Excipients did not interfere with the active moiety. The technique demonstrated good linearity, durability, accuracy, and precision. Omeprazole in bulk and its prescription dose forms were routinely analyzed using the suggested method.

#### ACKNOWLEDGEMENT

The authors are thankful to management and principal of Institute of Pharmaceutical Science and Research for Girls, Bhigwan-413130, Maharashtra for extending support to carry out the present research work.

#### REFERENCES

1. Mohan Prasad VG et al. Efficacy and Safety of Omeprazole for the Treatment of Acid Peptic Disorders: A Systematic Review and Meta-Analysis. International Journal of Clinical Practice, 2024: 1-13.
2. Massoomi, Fetal. Omeprazole: a comprehensive review Pharmacotherapy. Pharmacotherapy, 1993; 13(1): 46-59.
3. Gandhimathi, R et al. Analytical process of drugs by Ultra violet (UV) Spectroscopy –A review. Int Jour of Pharma Res & Anal, 2012; 2(2): 72-78.

4. Rutuja S Shah et al. UV-Visible spectroscopy –A review. *Int Jour of Ins Phar and Life Sci*, 2015; 5(5): 490-505.
5. Salomi Patta et al. Simultaneous estimation of aspirin and Omeprazole (yosprala) in bulk by UV spectroscopy. *Journal of Drug Delivery & Therapeutics*, 2017; 7(3): 87-91.
6. Marcelo Ribani et al. Validation of chromatographic methods: Evaluation of detection and quantification limits in the determination of impurities in omeprazole. *Journal of Chromatography A*, 2007; 1156: 201–205.
7. Vishwanath V et al. Method development and validation of Omeprazole by using UV spectroscopy. *Int Jour of Tren in Phar and Life Sci*, 2017; 3(5): 75-79.
8. Rakesh Singh, Swarnlata Saraf. Spectrophotometric Estimation of Omeprazole in Pharmaceutical Dosage Form. *Research J. Pharm. and Tech*, 2008; 1(3): 276-277.
9. Abdalla Ahmed Elbashir, Iman Ahmed Alamin. A new study on Omeprazole spectrophotometric determination using 9- Fluorenylmethyl chloroformate as derivatizing agent. *J Anal Pharm Res*, 2019; 8(2): 38-43.
10. Sudhakar Rao et al. Development and validation of UV spectroscopic method for the estimation of Omeprazole in bulk and pharmaceutical formulation. *Int J Pharm*, 2014; 4(1): 247-251.
11. Nuran Ozaltın, Aysegül Kocer. Determination of Omeprazole in pharmaceuticals by derivative spectroscopy. *Jour of Pharm & Biomed Anal*, 1997; 16(2): 337-342.
12. Riedel A. Development and validation of new analytical method for the simultaneous estimation of omeprazole and domperidone in pharmaceutical dosage form by UV spectrophotometry. *Int Jour of Res in Pharm Chem & Anal*, 2019; 1(2): 44-46.
13. Kumaraswamy D, Stephen Rathinaraj B, Rajveer CH, Sudharshini S, Bhupendra Shrestha, Rajasridhar. Statistical assurance of process validation by analytical method development and validation for omeprazole capsules and blend. *Res Jour of Pharma, Bio and Chem Sci*, 2010; 1(3): 50-54.
14. Harrizul Rivail et al. Development and Validation of Omeprazole Analysis Methods in Capsules with Absorbance Methods and Areas under Curves Methods with UV-Vis Spectrophotometry. *Int. Jour of Phar Sci & Med*, 2018; 3(3): 21-32.
15. Prashanthi Malyala et al. Validated UV Spectrophotometric Methods for the Simultaneous Estimation of Omeprazole and Domperidone. *Indian Journal of Advances in Chemical Science*, 2021; 9(3): 236-242.
16. Biju V M et al. Stability Indicating UV Spectrophotometric Method for Estimation of Omeprazole and Its Application to Content Uniformity Testing. *International Journal of Pharmaceutical Quality Assurance*, 2018; 9(2): 158-162.
17. Amol Bhandage et al. Extractive Spectrophotometric Determination of Omeprazole in Pharmaceutical Preparations. *Tropical Journal of Pharmaceutical Research*, 2009; 8 (5): 449-454.
18. Abdel-Aziz M Wahbi et al. Spectrophotometric determination of Omeprazole, lansoprazole and pantoprazole in pharmaceutical formulations. *Journal of Pharmaceutical and Biomedical Analysis*, 2002; 30(4): 1133-1142.
19. Pandey V.P S. Vanjinathan, Stella Singh. Determination of Domperidone and Omeprazole Capsule Dosage form by UV/Visible Spectrophotometer. *Asian Journal of Chemistry*, 2010; 22(3): 2305-2308.
20. Sandip S Chaudhari, Swapnil D Phalak. Development and validation of UV Spectrophotometric method for simultaneous equation of Aspirin and Omeprazole in tablet dosage form. *Pharmaceutica Analytica Acta*, 2020; 11(1): 1-5.
21. Varsha Balkrishna Mane et al, Development of UV Spectrophotometric Method for the Simultaneous Estimation of Domperidone and Omeprazole in Capsule Dosage Form by Simultaneous Equation and Absorbance Ratio

Method. Asian J. Research Chem, 2011; 4(7): 1119 -1124.

22. Claudia Setal. Quantification of omeprazole degradation by enteric coating polymers:an UV-Vis spectroscopy study. Pharmazie, 2005; 60: 126–130.
23. Vivek Jain,Neetesh K.Jain. Analytical Method Development and Validation for simultaneous determination of Amoxicillin,Omeprazole and Rifabutin in Bulk and in a synthetic mixture by UV spectroscopy. Journal of Drug Delivery and Therapeutics, 2019; 9(4) :788-794.
24. Pavan Kumar, V et al. Development and validation of new analytical method for the simultaneous estimation of omeprazole and domperidone in pharmaceutical dosage form by UV spectrophotometry. Int. J. Res. Pharm. Chem & Analy, 2019; 1(2): 44-46.
25. International Conference on Harmonisation of Technical Requirement of Registration of Pharmaceuticals for Human Use. Validation of Analytical Procedures: Text and Methodology, Q2B; 1996. Geneva, Switzerland.
26. Validation of analytical procedures: Text and Methodology Q2(R1)  
<https://database.ich.org/sites/default/files/Q2%28R1%29%20Guideline.pdf>