

REVIEW ON SOLUBILITY ENHANCEMENT TECHNIQUES IN PHARMACEUTICAL FORMULATIONS

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Article Received: 30 December 2025 | Article Revised: 20 January 2026 | Article Accepted: 9 February 2026

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DOI: <https://doi.org/10.5281/zenodo.18639613>

How to cite this Article: Vare Nithin Reddy, Dr V. Sai Kishore, Ch. Bhargavi, Mangalapudi Rami Reddy, Gajavilli Chandini, P. Saranya (2026) REVIEW ON SOLUBILITY ENHANCEMENT TECHNIQUES IN PHARMACEUTICAL FORMULATIONS. World Journal of Pharmaceutical Science and Research, 5(2), 574-581. <https://doi.org/10.5281/zenodo.18639613>



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ABSTRACT

Poor aqueous solubility is a major challenge in pharmaceutical formulation development, as it directly affects drug dissolution, absorption, and bioavailability. A significant proportion of newly developed drug molecules exhibit low water solubility, leading to formulation difficulties and reduced therapeutic effectiveness. To address this issue, a wide range of solubility enhancement techniques have been explored and implemented in pharmaceutical sciences. This review provides a comprehensive overview of conventional, advanced, and emerging approaches used to enhance the solubility of poorly water-soluble drugs in pharmaceutical formulations. Various physical, chemical, and formulation-based strategies such as particle size reduction, solid dispersion, salt formation, pH adjustment, complexation, surfactant-based systems, lipid-based drug delivery systems, nanotechnology-based approaches, and other novel techniques are discussed. Additionally, alternative and supportive approaches including hydrotropy, amorphization, electrospinning, and the use of solubilizing excipients are highlighted. Understanding the principles, advantages, and limitations of these techniques is essential for selecting an appropriate solubility enhancement strategy and for improving the clinical performance of poorly soluble drugs. This review aims to serve as a valuable reference for researchers, formulators, and students involved in pharmaceutical formulation development.

KEYWORDS: Solubility enhancement; Poorly water-soluble drugs; Bioavailability; Solid dispersion; Nanotechnology; Lipid-based drug delivery systems; Pharmaceutical formulations.

INTRODUCTION

Solubility is a fundamental physicochemical property that critically influences the performance of pharmaceutical formulations, particularly oral dosage forms, as drugs must be dissolved in aqueous environments to be absorbed and

exert their therapeutic effect. Drugs with low aqueous solubility often exhibit slow dissolution rates, incomplete absorption, and consequently poor bioavailability, presenting a major challenge in drug development and formulation design.^[1] With the continual emergence of new chemical entities (NCEs), the proportion of poorly water-soluble drugs has increased significantly, leading to a higher risk of failure in the transition from discovery to clinical use due to inadequate solubility and resultant low bioavailability. This challenge not only affects pharmacokinetic parameters such as absorption and distribution but also complicates formulation strategies across varied dosage forms.^[2] To address these limitations, numerous solubility enhancement techniques have been developed and utilized in pharmaceutical formulation science. These strategies range from traditional physical and chemical methods — such as particle size reduction, pH alteration, and salt formation — to more advanced approaches including solid dispersions, inclusion complexation with cyclodextrins, microemulsions, and nanotechnology-based delivery systems. By enhancing the aqueous solubility and dissolution profiles, these techniques aim to improve systemic absorption and therapeutic efficacy of poorly soluble drugs.^[3] Given the increasing prevalence of poorly soluble drug candidates in modern pharmaceutical research, a comprehensive understanding of solubility enhancement methodologies and their mechanisms is essential to optimize formulation design and improve clinical outcomes.^[4]

Methods for Enhancing Solubility

Poor aqueous solubility remains one of the major challenges in the development of effective pharmaceutical formulations, as it directly affects drug dissolution, absorption, and bioavailability. To overcome this limitation, a wide range of solubility enhancement techniques have been developed and successfully applied in formulation science. These methods aim to modify either the physical characteristics of the drug, its chemical environment, or the formulation approach to improve interaction with biological fluids.

One of the most employed approaches for solubility enhancement involves **physical modification of drug particles**. Reduction in particle size, achieved through techniques such as micronization and nanonization, increases the surface area available for dissolution, thereby enhancing the dissolution rate. Similarly, solid dispersion techniques, where the drug is dispersed in a hydrophilic carrier, improve wettability and reduce crystallinity, leading to enhanced solubility. Advanced physical processing methods such as hot-melt extrusion and cryogenic processing have also gained importance due to their ability to produce amorphous or molecularly dispersed drug forms with improved dissolution behavior.

Chemical methods play a significant role in improving solubility by altering the chemical nature or microenvironment of the drug. Salt formation is widely used for ionizable drugs, as it enhances solubility without affecting pharmacological activity. Adjustment of pH and use of co-solvents are simple yet effective techniques that increase drug solubility by improving ionization or reducing intermolecular forces. Co-crystallization has emerged as a promising alternative, allowing modification of solubility and stability through interaction with suitable co-formers while maintaining the drug's chemical identity.

Another important strategy for solubility enhancement is **complexation**, particularly inclusion complex formation with cyclodextrins. These cyclic oligosaccharides encapsulate hydrophobic drug molecules within their cavity, resulting in improved aqueous solubility and stability. Ion-pairing techniques are also employed to enhance solubility and membrane permeability, especially for charged drug molecules.

Surfactant-based and lipid-based formulation approaches

They have gained increasing attention due to their effectiveness in solubilizing poorly water-soluble drugs. Surfactants enhance solubility by reducing interfacial tension and forming micelles that entrap drug molecules. Self-emulsifying drug delivery systems, microemulsions, and nanoemulsions spontaneously form fine dispersions in gastrointestinal fluids, providing a large surface area for drug dissolution and absorption.

In recent years, **novel and advanced drug delivery systems** have significantly contributed to solubility enhancement. Nanoparticles, nanosuspensions, polymeric micelles, and liposomes improve solubility by reducing particle size to the nanometer range or by encapsulating hydrophobic drugs within carrier systems. These approaches not only enhance solubility but also offer additional benefits such as controlled drug release and improved therapeutic efficacy.

Overall, the selection of an appropriate solubility enhancement technique depends on the physicochemical properties of the drug, the intended dosage form, and the desired therapeutic outcome. A comprehensive understanding of these methods is essential for the successful formulation of poorly soluble drugs and for improving their clinical performance.

1. Particle Size Reduction

Particle size reduction is one of the simplest and most widely used techniques to enhance solubility. By reducing drug particles to micron or nanometer size through processes such as milling, micronization, or nanonization, the surface area available for dissolution is significantly increased. According to the Noyes–Whitney equation, an increase in surface area results in an enhanced dissolution rate, which ultimately improves bioavailability. This method is particularly effective for poorly soluble drugs intended for oral administration.

2. Solid Dispersion Technique

Solid dispersion involves dispersing a poorly soluble drug in a hydrophilic carrier such as polyethylene glycol (PEG), polyvinylpyrrolidone (PVP), or hydroxypropyl methylcellulose (HPMC). The drug may exist in an amorphous or molecularly dispersed form, which reduces crystallinity and improves wettability. As a result, solid dispersions significantly enhance dissolution rate and solubility compared to pure crystalline drugs.

3. Salt Formation

Salt formation is a commonly applied chemical method for enhancing solubility of ionizable drugs. In this approach, the drug is converted into a salt form by reacting it with suitable acids or bases. Salt forms generally exhibit higher aqueous solubility, faster dissolution, and improved stability. This technique is widely used in pharmaceutical industries due to its simplicity and effectiveness.

4. pH Adjustment

The solubility of weakly acidic or basic drugs depends strongly on the pH of the surrounding medium. pH adjustment improves solubility by increasing the ionization of the drug in solution. Buffer systems are often incorporated into formulations to maintain the desired pH, ensuring enhanced dissolution and absorption of the drug in the gastrointestinal tract.

5. Use of Co-solvents

Co-solvency involves the addition of water-miscible organic solvents such as ethanol, propylene glycol, or polyethylene glycol to increase the solubility of poorly soluble drugs. Co-solvents reduce the polarity of the aqueous system, thereby enhancing drug solubility. This method is commonly used in liquid formulations such as injections and oral solutions.

6. Complexation (Cyclodextrin Inclusion Complexes)

Complexation improves solubility by forming reversible complexes between the drug and complexing agents such as cyclodextrins. The hydrophobic drug molecule is encapsulated within the hydrophobic cavity of cyclodextrins, while the hydrophilic outer surface interacts with water. This leads to improved solubility, stability, and bioavailability of the drug.

7. Surfactant-Based Solubilization

Surfactants enhance solubility by reducing interfacial tension and improving wettability of drug particles. Above their critical micelle concentration, surfactants form micelles that entrap poorly soluble drugs within their hydrophobic core. This method is widely used in oral, topical, and parenteral formulations.

8. Self-Emulsifying Drug Delivery Systems (SEDDS)

SEDDS are isotropic mixtures of oils, surfactants, and co-surfactants that spontaneously form fine emulsions upon contact with gastrointestinal fluids. These systems maintain the drug in a solubilized state, enhance dissolution, and improve absorption, especially for lipophilic drugs with poor aqueous solubility.

9. Nanotechnology-Based Approaches

Nanoparticles and nanosuspensions enhance solubility by reducing drug particle size to the nanometer range, thereby increasing surface area and saturation solubility. These systems also improve drug stability and provide controlled release. Nanotechnology-based formulations are increasingly used for poorly soluble drugs with low bioavailability.

10. Lipid-Based Drug Delivery Systems

Lipid-based systems such as liposomes, microemulsions, and nanoemulsions enhance solubility by encapsulating poorly soluble drugs within lipid carriers. These systems improve drug dissolution, intestinal absorption, and bioavailability. Additionally, lipid-based formulations can reduce drug degradation and enhance therapeutic effectiveness.

11. Co-crystallization

Co-crystallization is an emerging solubility enhancement technique in which the drug is crystallized with a pharmaceutically acceptable co-former through non-covalent interactions such as hydrogen bonding. This method modifies the crystal lattice of the drug without altering its chemical structure, resulting in improved solubility, dissolution rate, and sometimes stability. Co-crystals are especially useful for drugs where salt formation is not feasible.

12. Amorphization

In this approach, the drug is converted from its crystalline form into an amorphous form, which possesses higher free energy and greater molecular mobility. As amorphous drugs lack long-range molecular order, they dissolve more

rapidly than crystalline forms. However, physical instability and recrystallization remain major challenges associated with this method.

13. Hot-Melt Extrusion (HME)

Hot-melt extrusion involves melting a drug with suitable polymers and forcing the mixture through an extruder to form a solid dispersion. This technique improves solubility by reducing crystallinity and ensuring uniform drug distribution at the molecular level. HME is a solvent-free, continuous process and is widely used in industrial-scale manufacturing.

14. Cryogenic Processing

Cryogenic techniques involve rapid freezing of drug solutions using cryogenic liquids such as liquid nitrogen. This process produces highly porous, amorphous, or fine particulate drug forms with improved wettability and dissolution rate. Cryogenic processing is particularly effective for thermolabile and poorly soluble drugs.

15. Use of Porous Carriers

Porous carriers such as mesoporous silica enhance solubility by adsorbing the drug into their pores. The drug remains in an amorphous or molecularly dispersed state, leading to increased surface area and rapid dissolution. This approach also improves drug stability and prevents recrystallization.

16. Supercritical Fluid Technology

Supercritical fluids, particularly supercritical carbon dioxide, are used to reduce particle size and modify drug morphology. This technique produces fine particles with narrow size distribution and improved solubility. Supercritical fluid technology is considered environmentally friendly as it reduces the use of organic solvents.

17. Polymeric Micelles

Polymeric micelles are formed from amphiphilic block copolymers that self-assemble in aqueous media. Poorly soluble drugs are entrapped within the hydrophobic core, resulting in enhanced solubility and stability. This method is particularly useful for delivering lipophilic drugs and improving bioavailability.

18. Prodrug Approach

The prodrug strategy involves chemical modification of the drug molecule to form a more soluble derivative that converts back to the active drug after administration. This method enhances solubility, permeability, and sometimes taste masking, although it requires extensive chemical and regulatory evaluation.

Other approaches for Enhancing Solubility

1. Hydrotropy

Hydrotropy involves the addition of large amounts of hydrotropic agents such as sodium benzoate, sodium citrate, or urea to enhance the solubility of poorly soluble drugs. Hydrotropes improve solubility by weakening intermolecular interactions within the drug and increasing its affinity toward the aqueous medium. This approach is simple, cost-effective, and particularly useful in analytical and liquid formulations.

2. Crystal Habit Modification

Crystal habit modification alters the external shape of drug crystals without changing their internal structure. By controlling crystallization conditions such as solvent, temperature, and additives, the surface properties of crystals can

be modified, leading to improved wettability and dissolution behavior. This method is advantageous because it does not require chemical modification of the drug.

3. Use of Solubilizing Excipients

Certain excipients act as solubilizers by improving wettability and dispersibility of drug particles. Polymers such as poloxamers, PVP, HPMC, and Soluplus® enhance solubility by forming molecular interactions with the drug and preventing aggregation. These excipients are widely used due to their safety and regulatory acceptance.

4. Drug–Polymer Complex Formation

Drug–polymer complexes are formed through hydrogen bonding or electrostatic interactions between the drug and polymer chains. These complexes improve solubility by stabilizing the drug in a dissolved or amorphous state. Polymers such as chitosan and polyacrylic acid are commonly used for this purpose.

5. Use of Ionic Liquids

Ionic liquids are salts that remain liquid at or near room temperature and can dissolve a wide range of poorly soluble drugs. They enhance solubility by disrupting strong crystal lattice forces. Although promising, their pharmaceutical application is still under investigation due to safety and toxicity concerns.

6. Foam Drying

Foam drying is a novel technique where drug solutions are converted into foams and rapidly dried to obtain highly porous structures. These porous matrices enhance surface area and wettability, resulting in rapid dissolution and improved solubility. This method is especially useful for thermosensitive drugs.

7. Freeze Drying (Lyophilization)

Lyophilization involves freezing the drug solution followed by sublimation of the solvent under vacuum. This process yields highly porous, amorphous solids with enhanced dissolution rates. Freeze drying is commonly used for injectable and rapidly dissolving oral dosage forms.

8. pH-Modulated Solid Dosage Forms

This approach incorporates pH modifiers directly into solid dosage forms to create a localized pH environment that enhances drug solubility at the site of dissolution. It is particularly effective for weakly basic or acidic drugs with pH-dependent solubility.

9. Electrospinning

Electrospinning produces ultra-fine drug-loaded fibers with very high surface area. Drugs incorporated into electrospun fibers often exist in an amorphous state, leading to rapid dissolution and enhanced solubility. This technique is gaining attention for oral and transmucosal drug delivery.

10. Use of Bio-enhancers

Bio-enhancers such as piperine and certain surfactants increase solubility and absorption by modifying membrane permeability and inhibiting drug metabolism. Although primarily used to enhance bioavailability, they indirectly support solubility improvement.

CONCLUSION

Solubility enhancement remains a critical aspect of pharmaceutical formulation development, particularly in view of the increasing number of poorly water-soluble drug candidates emerging from modern drug discovery programs. Inadequate solubility can significantly limit drug dissolution, absorption, and bioavailability, ultimately affecting therapeutic efficacy. Over the years, numerous solubility enhancement techniques have been developed, ranging from simple physical and chemical methods to advanced formulation and nanotechnology-based approaches.

This review highlights a broad spectrum of solubility enhancement strategies, including particle size reduction, solid dispersion, salt formation, pH modification, complexation, surfactant-based systems, lipid-based formulations, nanocarriers, and other innovative techniques such as co-crystallization, amorphization, supercritical fluid technology, and electrospinning. Each method offers distinct advantages and limitations, and no single approach is universally applicable to all drugs. Therefore, the selection of an appropriate technique should be based on the physicochemical properties of the drug, the intended dosage form, manufacturing feasibility, and regulatory considerations.

A thorough understanding of these solubility enhancement approaches enables formulators to design effective and stable pharmaceutical products with improved bioavailability and therapeutic outcomes. Continued research and development in this area are expected to further advance formulation technologies and contribute to the successful delivery of poorly soluble drugs in clinical practice.

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